

UV-Vis Spectroscopy Analysis of Chlorophylls and Carotenoids in the leaves of *Carica Papaya L.* (Male and Female) by Using Various Extracting Solvents in Kachchh and Saurashtra region

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Received 01 July 2022; Accepted 14 July 2022

Abstract: Present investigation is performed on the extraction of photosynthetic pigments like chlorophyll-a, chlorophyll-b and carotenoids in *Carica Papaya L.* (Male and Female). by using different solvent like methanol, ethanol, diethyl ether, acetone and DMSO. Different trend is observed in extraction rate for chlorophylls and carotenoids. Methanol was shown to have the highest extraction rate of carotenoids and chlorophylls (chlorophylls a and b) in the male *Carica Papaya L.* species from the Kachchh region. 95% ethanol demonstrates the maximum extraction of chlorophylls (Chlorophyll a and b) and carotenoids for female *Carica Papaya L.* species from the Kachchh area. Dimethyl-sulphoxide (DMSO) exhibits the highest extraction of chlorophylls (Chlorophyll a and b) from the leaves of the male *Carica Papaya L.* Saurashtra species, while 95 % ethanol is the best solvent for carotenoids. The species chlorophylls (chlorophyll a and b) and carotenoids provide the greatest amount of extraction in diethyl-ether for female *Carica Papaya L.* from the Saurashtra region.

Key Word: Solvent extraction, chlorophylls, carotenoids, *Carica Papaya L.*, Kachchh, Saurashtra, UV-VIS Spectroscopy analysis

Abbreviations and notation: DMSO: Dimethyl-sulphoxide, DEE: Diethyl-ether

I. Introduction

More than just finding new medicines is studied in medicinal plant research.¹ Natural goods offer limitless potential for the novel medicine, whether they are used as pure components or standardized plant extracts.² This field has been growing and currently covers a wide range of topics, including power negotiation based on an understanding of medicinal plants.³ Arthro glycosides, bitter substances, flavonoids, alkaloids, saponins, coumarins, phenol carboxylic acids, and terpenes are only a few of the various phytoconstituents found in plants.⁴ These phytoconstituents endow plants with particular traits and abilities.⁵

Determining the diverse biological activities of plants would therefore benefit from the examination of their elements. For the qualitative and quantitative examination of pharmaceutical and biological materials, the spectroscopic approach has developed into a potent analytical tool.⁶ There is a need for simple, affordable, and quick assays to find phytocomponents.⁷ In this regard, both traditional procedures and spectroscopic (UV-Vis) methods can be applied. Ultraviolet-visible spectroscopy (UV-Vis) related to the quantitative method for calculating how much a chemical substance absorbs light is called spectroscopy (or spectrophotometry).⁸ This is accomplished by comparing the amount of light that passes through a sample to the amount of light that goes through a reference sample or a blank.⁹ The visible spectrum or its nearby wavelengths are used in UV-visible spectroscopy.¹⁰ The absorption in the visible ranges is directly influenced by the colour of the substances involved. In various portions of the electromagnetic spectrum, molecules go through electronic transitions.¹¹

Carica Papaya L. is an important plant species traditionally used in Indian system of medicine mainly for the treatment of various diseases such as asthma, arthritis, joints pain, ear ache, dog bite, chronic ulcer, fever, cough, cold, menstrual disorder, bone fracture, bronchitis, dysentery, dyspepsia, tumor, constipation, skin diseases, ascites and stomach distension.^{12,13,14} The main objective of the present study is to identify the phytoconstituents like chlorophyll-a, chlorophyll-b and carotenoids of *Carica Papaya L.* leaves extract by UV-VIS spectrum.

II. Material And Methods

Collection of plant sample:The plant *Carica Papaya L.* was collected from the region of Kachchh and Saurashtra, Gujarat, India. The authenticity of the plant was confirmed in Botanical Survey of India. The whole plant was washed under running tap water, shade dried at room temperature and powdered. Preparation of plant extract for UV-VIS spectrum analysis the shade dried whole plant of *Carica papaya L.* (at 25 °C) was powdered in a mechanical grinder. Analysed for the determination of chlorophylls (Ch-a and Ch-b) and carotenoids content.

Analytical procedure:Accurately weighted 0.5g of fresh plant leaves of *Carica Papaya L.* Sample was taken, and homogenized in tissue homogenizer with 10 ml of different extractant solvent. Homogenized sample mixture was centrifuge for 5000 rpm for 15min at 25°C. The supernatant were separated and 1 ml of it is mixed with 4 ml of the respective solvent. The solution mixture was analysed for Chlorophyll-a, Chlorophyll-b and carotenoids content in UV spectroscopy (shimadzu). The equation used for the quantification of Chlorophyll-a, Chlorophyll-b, and carotenoids by different extractant solvents are given in table 1; and spectral absorbance for Chlorophyll-a, Chlorophyll-b, and carotenoids for various solvents are represented in table.

Quality control:Analytical reagents used during the extraction process were of AR grade (Merck). Milli Q water was used for preparation of intermediate solution and for dilution purpose (wherever needed). Quartz cuvette (1cm²) was used and corresponding solvent was taken as reference during spectrophotometric observation. Every procedure (for each plant sample and extracting solvent) was triplicated for maintaining the precision of analytical results.

Table-1 Equations to determine concentrations (µg/ml) of chlorophyll a (Ch-a), chlorophyll b (Ch-b) and total carotenoids (C x+c) by different extractant solvents in UV spectroscopy^{15, 16, 17}

Solvents	Formula
95% Ethanol	$Ch-a=13.36A_{664} - 5.19 A_{649}$ $Ch-b=27.43A_{649} - 8.12 A_{664}$ $C x+c=(1000A_{470} - 2.13Ca - 97.63Cb)/209$
Methanol	$Ch-a=16.72A_{665.2} - 9.16A_{652.4}$ $Ch-b=34.09A_{652.4} - 15.28A_{665.2}$ $C x+c=(1000A_{470} - 1.63Ca - 104.96Cb)/221$
80% Acetone	$Ch-a=12.25A_{663.2} - 2.79A_{646.8}$ $Ch-b=21.5A_{646.8} - 5.1A_{663.2}$ $C x+c=(1000A_{470} - 1.82Ca - 85.02Cb)/198$
Diethyl-ether (DEE)	$Ch-a=10.05A_{660.6} - 0.97A_{642.2}$ $Ch-b=16.36A_{642.2} - 2.43A_{660.6}$ $C x+c=(1000A_{470} - 1.43Ca - 35.87Cb)/205$
Dimethyl-sulphoxide (DMSO)	$Ch-a=12.47A_{665.1} - 3.62A_{649.1}$ $Ch-b=25.06A_{649.1} - 6.5A_{665.1}$ $C x+c=(1000A_{480} - 1.29Ca - 53.78Cb)/220$

A = Absorbance, Ch-a = Chlorophyll a, Ch-b = Chlorophyll b, C x+c = Carotenoids

III. RESULTS AND DISCUSSION

The qualitative UV-VIS spectrum profile of solvent extract of *Carica Papaya L.* was selected at a wavelength from 190 to 800 nm due to the sharpness of the peaks and proper baseline. The profile showed the peaks absorbance respectively.

The primary pigments that transform light energy into chemical energy are known as chlorophyll-a. By transmitting the light it absorbs to chlorophyll-a, chlorophyll-b, an auxiliary pigment, indirectly contributes to photosynthesis.¹⁸The chlorophyll molecule has an ionic and hydrophilic center made of Mg²⁺, and a hydrophobic ring with a carbonyl group at its tail that is polar in nature.¹⁹ Water-soluble chlorophyll-binding protein keeps it in place within the plant cell (WSCP). Only one functional group (i.e., -CHO) attached to the porphyrin ring distinguishes chlorophyll-b from chlorophyll-a, and because of this, chlorophyll-b is more soluble than chlorophyll-a in polar solvents.¹⁷

Structure: Chemdraw structure of Chlorophyll-a and Chlorophyll-b

Structure: Chemdraw structure of Carotenoids

Table-2 UV spectroscopic determination of absorbance for Chlorophyll a, Chlorophyll b and Carotenoids with various extracting solvents.

Extractant solvent	Carica papaya male leaves kachchh	Carica papaya female leaves kachchh	Carica papaya male leaves Saurashtra	Carica papaya female leaves Saurashtra
95 % Ethanol				
A664nm Ch-a	0.526	0.647	0.654	0.711
A649nm Ch-b	0.279	0.362	0.341	0.369
A470 nm C x+c	0.690	0.890	1.023	0.903
Methanol				
A665.2 nm Ch-a	0.652	0.299	0.387	0.169
A652.4nm Ch-b	0.417	0.192	0.223	0.108
A470 nm C x+c	1.023	0.434	0.447	0.225
80 % Acetone				
A663.2 nm Ch-a	0.584	0.531	0.452	0.763
A646.8 nm Ch-b	0.291	0.295	0.204	0.398
A470 nm C x+c	0.721	0.704	0.552	0.890

Diethyle ether				
A660.6 nm Ch-a	0.509	0.457	0.320	0.926
A642.2 nm Ch-b	0.176	0.209	0.132	0.488
A470 nm C x+c	0.468	0.534	0.384	1.909
Dimethyl sulfoxide				
A665.1 nm Ch-a	0.387	0.215	0.711	0.464
A649.1 nm Ch-b	0.186	0.124	0.369	0.224
A480 nm C x+c	0.360	0.218	0.669	0.451

Table-3 Quantification of Chlorophyll a, Chlorophyll b and Carotenoids ($\mu\text{g/ ml}$) of *Carica Papaya L.* in various chemical solvents

Plants	Male leaves Kachchh	Female leaves kachchh	Male leaves Saurashtra	Female leaves Saurashtra
Time	15 min	15 min	15 min	15 min
95 % Ethanol				
Ch-a	5.58	6.76	6.97	7.58
Ch-b	2.40	4.68	4.04	4.35
C x+c	2.12	2.00	2.93	2.21
Methanol				
Ch-a	7.08	3.23	4.43	1.83
Ch-b	4.25	1.97	1.69	1.1
C x+c	2.56	1.00	1.19	0.48
80 % Acetone				
Ch-a	6.34	5.68	4.96	8.24
Ch-b	3.28	3.63	2.08	4.66
C x+c	2.19	1.94	1.85	2.42
Diethyle ether				
Ch-a	4.95	4.39	3.09	8.84
Ch-b	1.63	2.3	1.38	5.73
C x+c	1.96	2.17	1.61	8.25
Dimethyl sulfoxide				
Ch-a	4.15	2.23	7.54	4.98
Ch-b	2.14	1.72	4.63	2.59
C x+c	1.09	0.56	1.86	0.82

Methanol was shown to have the highest extraction rate of carotenoids and chlorophylls (chlorophylls a and b) in the male *Carica Papaya L.* species from the Kachchh region. 95% ethanol demonstrates the maximum

extraction of chlorophylls (Chlorophyll a and b) and carotenoids for female *Carica Papaya L.* species from the kachchh area.

Dimethyl-sulphoxide (DMSO) exhibits the highest extraction of chlorophylls (Chlorophyll a and b) from the leaves of the male *Carica Papaya L.* Saurashtra species, while 95 %ethanol is the best solvent for carotenoids. The species chlorophylls (chlorophyll a and b) and carotenoids provide the greatest amount of extraction in diethyl-ether for female *Carica Papaya L.* from the Saurashtra region

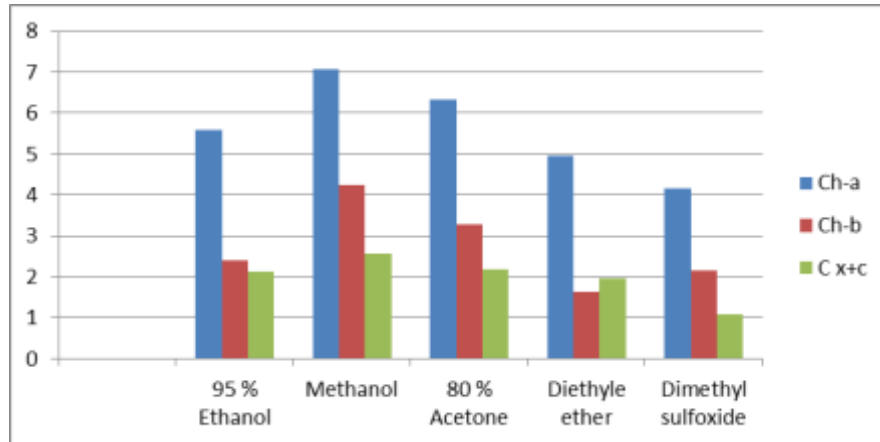


Fig. 1. The concentrations (µg/ ml) of Chlorophyll a, Chlorophyll b and Carotenoids in *Carica Papaya L.* Male leaves kachchh region

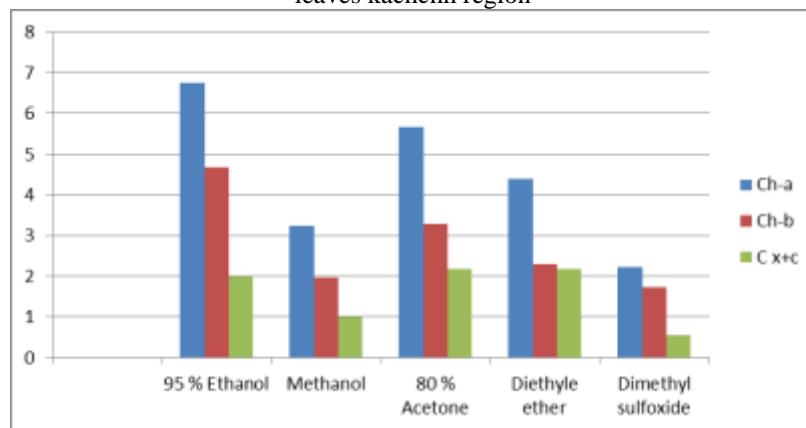


Fig. 2. The concentrations (µg/ ml) of Chlorophyll a, Chlorophyll b and Carotenoids in *Carica Papaya L.* Female leaves kachchh region

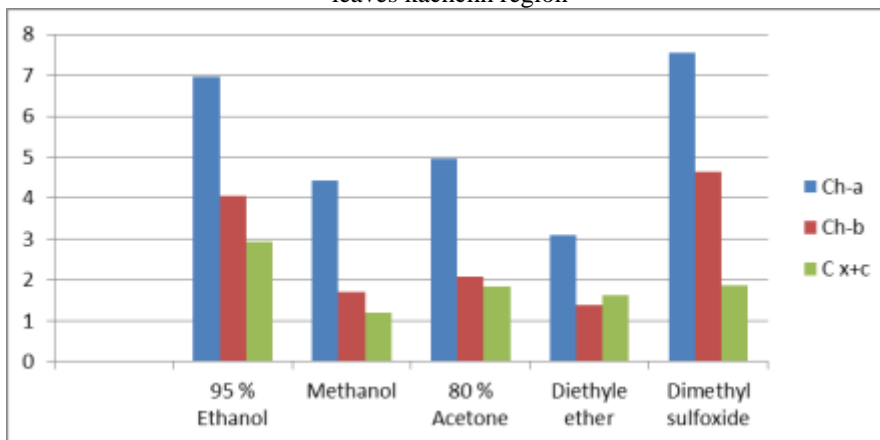


Fig. 3. The concentrations (µg/ml) of Chlorophyll a, Chlorophyll b and Carotenoids in *Carica Papaya L.* Male leaves Saurashtra region

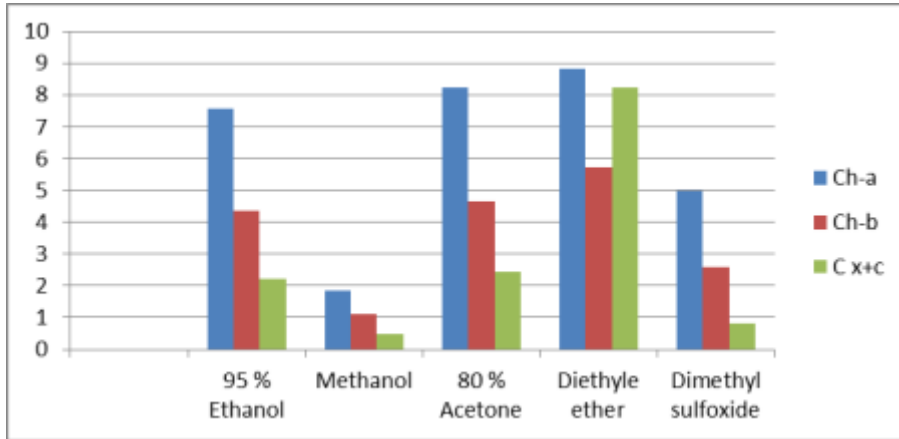


Fig. 4 The concentrations (µg/ml) of Chlorophyll a, Chlorophyll b and Carotenoids in *Carica Papaya L.* Female leaves Saurashtra region

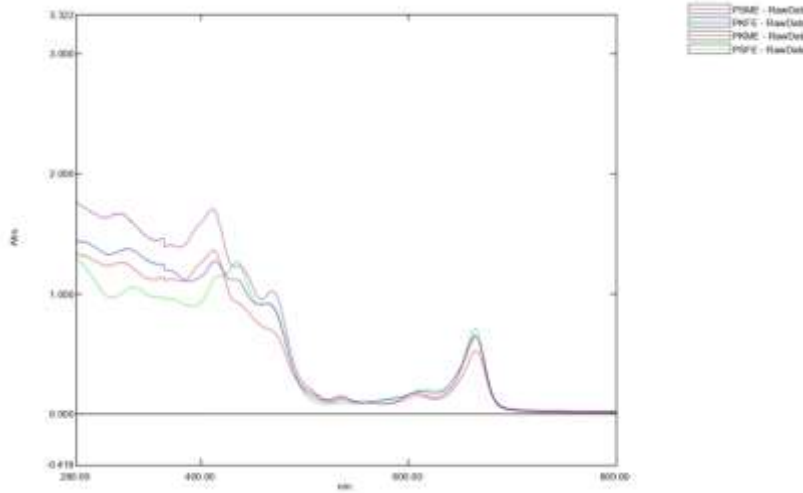


Fig. 5. UV spectroscopic determination of 95% ethanol

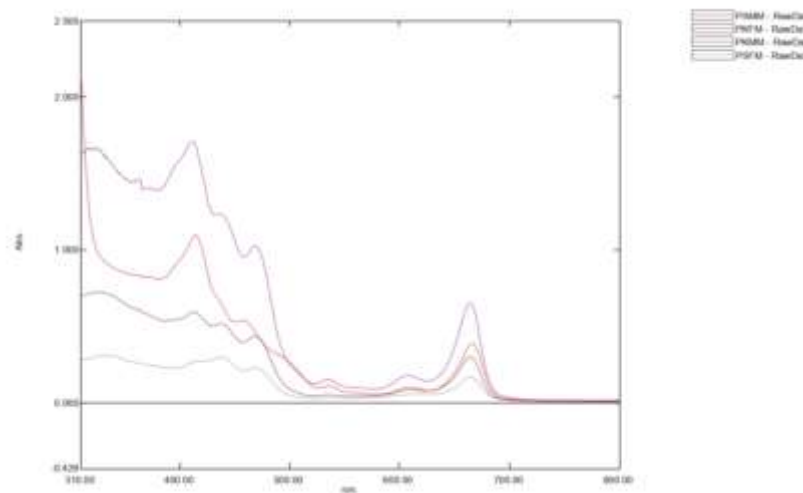


Fig.6. UV spectroscopic determination of methanol

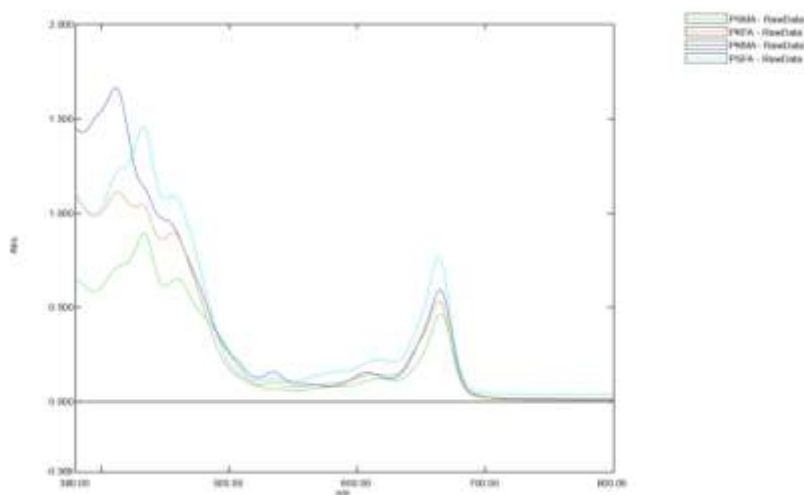


Fig.7. UV spectroscopic determination of 80% Acetone

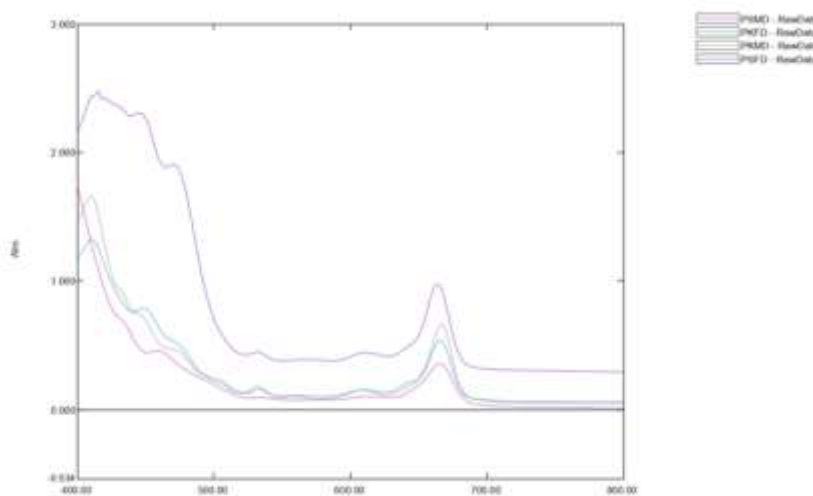


Fig.8 UV spectroscopic determination of Diethyle ether

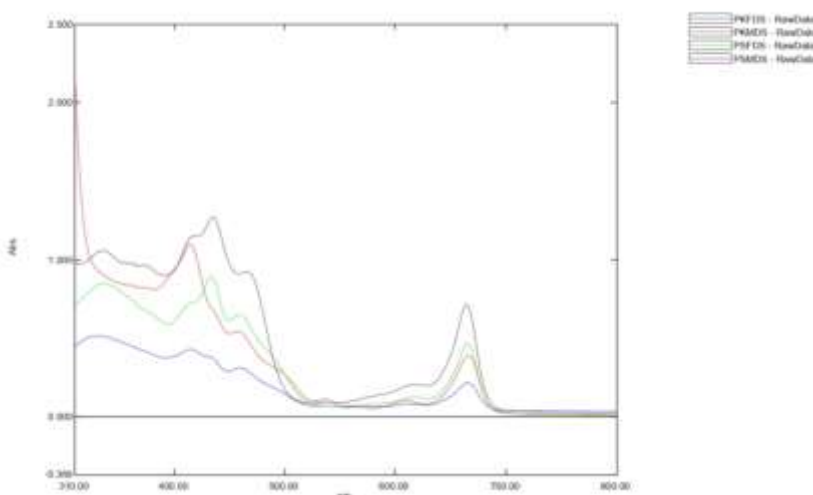


Fig.9 UV spectroscopic determination of Dimethyl-sulphoxide

IV. Discussion

The extractions of chlorophylls and carotenoids by different solvent for examined species are in the sequence of –

Male *Carica Papaya L. kachchh region* –

Chlorophyll-a: Methanol > 80% Acetone > 95% Ethanol > DEE > DMSO. **Chlorophyll-b:** Methanol > 80% Acetone > 95% Ethanol > DEE > DMSO. **Carotenoids:** Methanol > 80% Acetone > 95% Ethanol > DEE > DMSO

Female *Carica Papaya L. kachchh region* –

Chlorophyll-a: 95% Ethanol > 80% Acetone > DEE > Methanol > DMSO **Chlorophyll-b:** 95% Ethanol > 80% Acetone > DEE > Methanol > DMSO **Carotenoids:** DEE > 95% Ethanol > 80% Acetone > Methanol > DMSO

Male *Carica Papaya L. Saurashtra region* –

Chlorophyll-a: DEE > 95% Ethanol > 80% Acetone > Methanol > DMSO. **Chlorophyll-b:** DMSO > 95% Ethanol > 80% Acetone > Methanol > DEE **Carotenoids:** 95% Ethanol DMSO > 80% Acetone > DEE > Methanol

Female *Carica Papaya L. kachchh region* –

Chlorophyll-a: DEE > 80% Acetone > 95% Ethanol > DMSO > Methanol **Chlorophyll-b:** DEE > 80% Acetone > 95% Ethanol > DMSO > Methanol **Carotenoids:** DEE > 80% Acetone > 95% Ethanol > DMSO > Methanol

Carotenoids are found in chromoplasts, which give vegetables and fruits their colour, as well as in chlorophylls, which function in conjunction with chlorophylls in both photosystems. About 70 chemicals belong to the carotenoids group and their derivatives, which are found in the majority of plants and fruits.²⁰ According to Vechetel and Ruppel, carotene pigments are the most significant photosynthetic pigments because they protect chlorophyll and the thylakoid membrane from oxidative damage caused by ingested energy. The extraction of carotenoids using different solvents from the species under study shows a tendency that is remarkably similar to the one seen for chlorophylls

V. Conclusion

The findings of this study demonstrate unequivocally that the chemical composition of biomolecules determines which solvents are best for extracting photosynthetic pigments (chlorophyll-a, chlorophyll-b and carotenoids). According to research, methanol is the best extraction solvent for extracting the largest amount of Ch-a, Ch-b, and carotenoids from male *Carica Papaya L.* leaves from the Kachchh region. In the ethanol solvent, the female leaves of *Carica Papaya L.* from the Kachchh region exhibit significant levels of Ch-a, Ch-b, and carotenoids. Male leaves of the *Carica Papaya L.* exhibit significant levels of Ch-a and Ch-b in DMSO solvent for the Saurashtra region sample. The highest levels of carotenoids are found in ethanol for the male *Carica Papaya L.* Diethyle ether is the best solvent and provides the most amount of extraction for female leaves in the Saurashtra region when it comes to Ch-a, Ch-b, and carotenoids. The present investigation found the greatest significant variations in the trend for pigment extraction in *Carica Papaya L.* reported before from phytoplankton and also from higher plants. Further research in this area is advised since variations in pigment concentrations in plants may potentially be caused by temporal and seasonal changes as well as local geological conditions.

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